

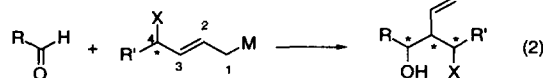
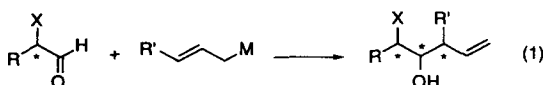
Synthesis of Chiral Allyltitaniums Having an Amino Group at the C-4 Position and Their Diastereoselective Addition Reaction with Aldehydes

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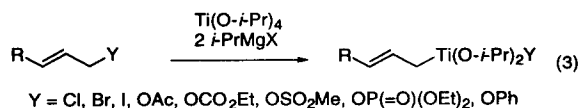
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Abstract: Chiral allyltitaniums having an amino substituent at the C-4 position are prepared from optically active allylic alcohol derivatives **1** and a $Ti(O-i-Pr)_4/2i-PrMgCl$ reagent, which, in turn, react with aldehydes regio- and stereoselectively to afford 3-amino-2-vinylalkanols in excellent yields. © 1997 Elsevier Science Ltd.

Reactions of substituted allylmagnesiums with aldehydes play an important role in stereoselective synthesis since, in one step, plural stereogenic centers can be generated and the resulting homoallyl alcohols can be used for further chemical transformations. Much effort has been paid to applying the reaction for the construction of three consecutive stereogenic centers. In consequence, there have been developed a variety of reactions of allylic metal compounds with chiral aldehydes as shown in eq 1 which proceed highly selectively with respect to the new stereocenters generated in concert with the new C-C bond formation (simple diastereoselectivity) as well as the stereocenter induced by the chiral center in the aldehyde (diastereofacial selectivity).¹ Diastereoselective addition reaction of achiral aldehydes with chiral allylmagnesiums possessing a stereocenter at the C-4 position as shown in eq 2 might afford another attractive methodology. However, investigations so far reported from the standpoint of application of the reaction for acyclic stereocontrol have been limited in regard to both the metal and the allylic moiety,² and further research is needed to gain recognition of the reaction as useful methodology like the reaction of eq 1. This restricted research work is presumably due to the difficulty of access to a variety of allylmagnesiums having a heteroatom substituent at C-4, the presence of which seems to be highly desirable for attaining excellent diastereoselectivity and also for further synthetic elaboration of the reaction products.³

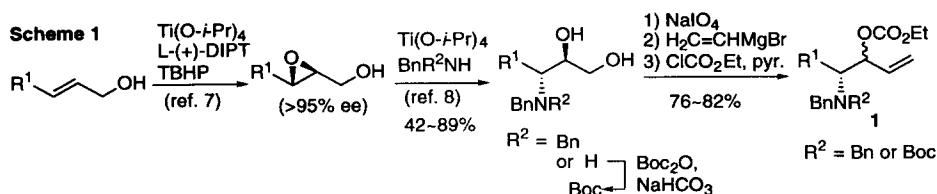


Recently, we have developed a new method for synthesizing allyltitanium complexes by the reaction of allyl halides or allyl alcohol derivatives with a $Ti(O-i-Pr)_4/2i-PrMgX$ reagent (eq. 3).⁴ Herein we report the



synthesis of allyltitaniums having an amino group at C-4 from allylic alcohol derivatives **1** containing an α -(1-amino)alkyl group and a Ti(O-*i*-Pr)₄/2*i*-PrMgX reagent and their highly selective reaction with an aldehyde.

Optically active substrates **1a**–**d**⁵ were readily prepared starting with primary allylic alcohols according to the conventional reaction sequence shown in Scheme 1. The carbonates **1** thus synthesized were treated with a Ti(O-*i*-Pr)₄/2*i*-PrMgX reagent and then with propanal and/or benzaldehyde. The stereoselectivity of the reaction was determined by GC, ¹H NMR and ¹³C NMR analysis of the crude hydrolysis products, while stereochemistries of the main product(s) were verified by ¹H NMR analysis after derivatization to the corresponding β -lactam and β -lactone by using the conventional reaction sequence.⁶ The results are summarized in Table 1.



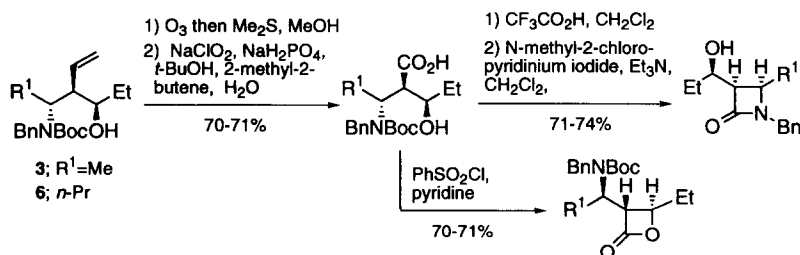
As can be seen from Table 1, allyltitanium compounds having an amino group at C-4 are readily prepared from **1** and a Ti(O-*i*-Pr)₄/2*i*-PrMgX reagent and, in turn, they react with an aldehyde to afford the expected γ -addition products in excellent yields. The diastereoselectivity of the reaction depends on whether R¹ in **1** is an alkyl or aryl and whether R³ in the aldehyde is an alkyl or aryl; however, in the case of another substituent R² on the N-atom the effect is not serious. Allyltitanium complexes derived from **1** where R¹ is an *alkyl* such as **1a**, **1b**, and **1c** react with an *alkyl aldehyde* to afford one product with the structure shown in Table 1 highly predominantly among four possible diastereomers (entries 1, 2 and 4). The total stereoselectivity (simple diastereoselectivity and diastereofacial selectivity) of more than 90% attained here is higher than the simple diastereoselectivity observed in the reaction of crotyltitanium and propanal which provides two possible diastereomers in a ratio of 75:25.⁴ In the case of either starting with **1** where R¹ is an aryl group (entry 5) or using an aryl aldehyde (entry 3), the diastereofacial selectivity induced by the stereogenic center at C-4 was very low. In these cases, although the stereochemistry with regard to the new C-C bond forming reaction was highly controlled, to our surprise, the sense of this simple diastereoselectivity was different between the reactions of entries 3 and 5.

The excellent stereoselectivity attained in the reaction of allylic titanium complexes derived from **1a**, **1b**, or **1c** with an alkyl aldehyde can be explained by assuming that the reaction proceeds *via* the six-membered chair-like transition state **a** (Fig. 1) which is most favorable according to the Felkin-Ahn model because the steric repulsion between the sterically most demanding amino moiety and the R³-group is minimized.⁹ In the case of entry 5, the reaction might proceed *via* the transition states **a** as well as **b** due to the small difference in the steric requirement between the amino and the phenyl (R¹) group, thus eventually resulting in low diastereofacial selectivity. The simple diastereoselectivity observed in entry 3 can be explained by assuming that severe gauche repulsion between both the sterically demanding phenyl group (R³) and the allyl substituent does

Table 1.^a

Entry	1		R ³ CHO	Products		
	R ¹	R ²		Main Diastereomer(s)	Ratio [main(s) : others]	Yield ^b %
	 $\xrightarrow[\text{ether, -50~-40 } ^\circ\text{C}]{\text{Ti(O-}i\text{Pr)}_4 / 2\text{-}i\text{PrMgCl}}$ $\xrightarrow{\text{R}^3\text{CHO}}$ 					
1	a ; Me	Bn	EtCHO	 2	[92 : 8] ^d	88
2	b ; Me	Boc	EtCHO	 3	[91 : 9] ^d	88
3	b ; Me	Boc	PhCHO	 4 5 45 : 55 ^f	[>95 : <5]	79
4	c ; <i>n</i> -Pr	Boc	EtCHO	 6	[91 : 9] ^d	89
5	d ; Ph	Boc	EtCHO	 7 8 58 : 42 ^g	[>95 : <5]	83

^aReaction procedure: To a solution of **1** (1.0 mmol) and $\text{Ti(O-}i\text{Pr)}_4$ (592 μL , 2.0 mmol) in ether (10 mL) was added $i\text{-PrMgCl}$ (1.0–1.5 M in ether, 4.0 mmol) at -50°C . After being stirred for 1.5 h at $-50 \sim -40^\circ\text{C}$, aldehyde (1.5 mmol) was added at -50°C and the resulting mixture was warmed to 0°C over 1 h followed by usual extractive workup. ^bTotal yield. ^cThe structure was confirmed by comparison with authentic **2** derived from **3** by treatment with $\text{CF}_3\text{CO}_2\text{H}$ and then BnBr . ^dThree diastereomers were detected on NMR and/or GC analysis (ratio = 92 : 2 : 6 for entry 1, 91 : 2 : 7 for entry 2, 91 : 3 : 6 for entry 4). ^eStereochemistry was confirmed by derivatization to the corresponding β -lactam and β -lactone according to the scheme shown below. ^fSimilar derivatization provided a *cis* and *trans* mixture of the corresponding β -lactam, while the β -lactones obtained have *cis*-configuration. ^gSimilarly, a mixture of *cis* and *trans* β -lactams were obtained. Treatment of the β -hydroxycarboxylic acid derived from **7** or **8** under the Mitsunobu reaction conditions (PPh_3 , DEAD, THF) afforded 1-amino-1-phenyl-2-pentene with (*Z*)-stereochemistry¹¹ (lactonization by using PhSO_2Cl /pyridine provided a complex mixture).



not allow the reaction to proceed *via* a transition state like **a** or **b**; the reaction thus proceeds *via* **c** in which the phenyl group occupies the axial position.¹⁰

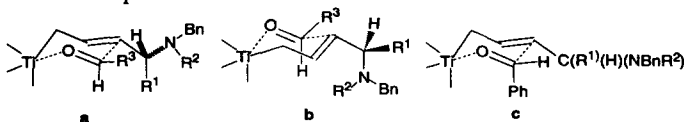


Fig. 1

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